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BUILDING MATERIAL PRODUCTS CONTAINING ORGANIC POLYMERS AS THICKENERS

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Building material products containing organic polymers as thickeners

Building material products, such as, for example, jointing compounds, adhesives, emulsion paints, synthetic plasters, fillers and the like, comprising a mixture of

- a) at least 10% by weight, in particular 10 to 99% by weight, of a water-soluble natural polymer and derivatives thereof,
- b) 1 to 89% by weight of an alkali metal salt or ammonium salt of a crosslinked polyacrylate which, if desired, is grafted with a starch, and
- c) 0 to 60t by weight of a water-soluble alkali metal salt or alkaline earth metal salt of arylsulfonic acid/formaldehyde condensation products or of a sulfo-modified melamine/formaldehyde polycondensation product.



ORIGINAL COMPLETE SPECIFICATION STANDARD PATENT

Application Number:

Lodged:

Invention Title:

BUILDING MATERIAL PRODUCTS CONTAINING ORGANIC POLYMERS

AS THICKENERS

Th following statement is a full description of this invention, including the b st method of performing it known to us

Building material products containing organic polymers as thickeners

- 5 It is known that derivatives of natural polymers, such as cellulose ethers, xanthan gums, guar derivatives are capable of increasing the viscosity in aqueous media to a substantial degree as a function of their chain length and concentration. The physical effects linked thereto 10 make it possible to utilize these substances as thickeners, water-retention agents, protective colloids, dispersants, stabilizers and binders in a large number of industrial applications. The variety of properties of these polymers enables their complex use in a wide range 15 of different industrial and product sectors. These include building materials, paints, adhesives, detergents cleaning compositions, cosmetics, foodstuffs, pharmaceuticals, products of the textile, leather, paper, and ceramics industries.
- The crucial and most important characteristic of the polymers is their viscosity in aqueous solution under specific conditions (concentrations, temperature, measuring device, shear gradient). For many applications, polymers are required which, when dissolved in water, give high viscosities, as a result of which a high thickening effect and water retention but also a significant improvement in stability can be achieved, ideally at relatively low concentrations of the polymers.
- Many measures aim at increasing the viscosity-increasing effect of the polymers and raising their economy and efficiency. Particularly careful selection of the starting raw materials (in the case of cellulose ethers: high-quality chemical pulps or cotton linters), in combination with chemical reactions avoiding excessive chain degradation (for example etherification) and

particular measures for gentle drying and fine grinding are the prerequisit fr maintaining a high degree of polymerization and preparing highly viscous macromolecular substances. Mor over, the chain length of the polymers can be lengthened by suitable chemical reactions, for example by crosslinking reactions with dichlorinated aromatics or olefins.

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For example, in the case of cellulose ethers the measures and developments mentioned lead to viscosity ranges (measured as a 2% aqueous solution) of more than 100,0000 mPa.s, in exceptional cases of up to 400,000 mPa.s. Despite fine grinding under conditions to give usable powdered products, nevertheless, chain degradation of the polymers, which in some cases can be significant, and which necessarily leads to reduced viscosities takes place. This involves economic disadvantages, since high-quality raw materials are expensive and a complicated process resulting in reduced yield is necessary for the preparation.

20 Furthermore, it is known that polyacrylates are used in the form of their alkali metal salts and ammonium salts as valuable thickeners for controlling the rheology of aqueous systems, for example in emulsion paints, pastelike fillers and adhesives. A special group are the 25 crosslinked polyacrylates or polyacrylates which are crosslinked and additionally grafted with starch, which are increasingly being used as so-called superabsorbents (SAPs) in hygiene articles (diapers, sanitary napkins), in agriculture and in horticulture, in the transport of 30 sensitive foodstuffs, in pharmaceutics, cosmetics, internal construction, as sludge solidifier and for sealing landfills.

The capacity of these polymers to absorb large volumes of aqueous liquids with the formation of a stable gel structure and even to withstand stress is the prerequisite for their successful use in the areas described above. It is

true that experimental applications of SAPs in the ar as f building mat rials, paints and adhesiv s show a thickening effect, which in some cases is clearly noticeable, despite their distinct properties t absorb x-traordinarily high amounts of aqueous liquids (up to 400 to 1000 times their own weight), but they do not even come close to reaching the high water retention capacity of natural polymers (such as, for example, of cellulose ethers) which is also necessary for the abovementioned systems.

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Furthermore, it is known that alkali metal salts and alkaline earth metal salts of the condensation products of arylsulfonic acids, i.e. of phenolsulfonic acids and naphthalenesulfonic acids, are used together with formal-dehyde as stabilization aid, dispersants, plasticizing agents and liquefying agents in various branches of industry. The condensation products mentioned are of particular importance as plasticizing and liquefying agents in concrete and mortar production in the building industry.

The high dispersability and liquefying effect lead in mortar and concrete, especially where a large amount is metered in, to extensive deposition of a water/binder slurry, as a result of which the initially present good plasticity and processability is substantially lost upon storage of the mortar or concrete, due to high internal compression.

A simple and low-cost method is desired which significantly increases the viscosity ranges of the natural high polymers in aqueous media and thus substantially improves the thickening effect.

Surprisingly, it has now been found that by using combinations of water-soluble natural polymers or derivatives thereof, such as cellulose ethers, xanthangums, guar derivatives, starch ethers, carobseed flour

with alkali metal salts and ammonium salts of crosslinked and optionally additionally starch-grafted polyacrylate superabsorb nts, and, if desired, with additions of alkali metal salts and alkaline earth metal salts of condensation products of naphthalenesulfonic acid or phenolsulfonic acid with formaldehyde or with sulfomodified melamine/formaldehyde polycondensation products, viscosity ranges in aqueous media can be achieved which are significantly higher than those of the starting substances mentioned. Combinations of the natural polymers mentioned or derivatives thereof superabsorbents alone produce a significant increase in the viscosity ranges compared with those of the starting substances used.

The invention relates to building material products comprising a mixture of

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- at least 10% by weight, in particular 10 to 99% by weight, preferably 40 to 90% by weight, of a water-soluble natural polymer and derivatives thereof,
- b) I to 89% by weight, preferably 5 to 60% by weight, of an alkali metal salt or ammonium salt of a crosslinked polyacrylate which, if desired, is additionally grafted with a starch, and
- of a water-soluble alkali metal salt or alkaline earth metal salt of arylsulfonic acid/formaldehyde condensation products or of a sulfo-modified melamine/formaldehyde polycondensation product.

The term "building material products" is here and hereinafter understood to mean fillers, adhesives, emulsion paints, synthetic plasters, jointing compositions and other products of this type.

In order to increase the viscosity, these products contain a mixture of the abovementioned components a) to c) in the composition given. These components will be described helow in more detail.

- a) Water-soluble natural polymers and derivatives th reof
 Preferred examples of these ar th following products:
- 1. Cellulose ethers

 The viscosity limits of these cellulose ethers are usually 10 500,000, in particular 50-150,000, mPa.s (measured as a 2% aqueous solution using a Höppler falling-ball viscometer at 20°C in distilled water). The most important cellulose ethers in practical application have the following etherification data:

			<u>DS</u>	MS
		Methylcellulose	1.4 - 2.3	
		Methylhydroxy thylcellul se		0.05 - 0.5
_		Methylhydroxypropylcellulose	1.3 - 2.2	0.1 - 1.0
5		Bydroxyethylcellulose	-	1.8 - 3.5
		Hydroxyethylhydroxypropyl-		
		cellulose	MS HE:	0.9 - 1.2
			MS HP:	0.6 - 0.9
10		Hydroxypropylcellulose	-	2 - 3.5
		Ethylhydroxyethylcellulose	0.7 - 1.2	0.8 - 2.7
		Carboxymethylcellulose	0.5 - 1.5	
		Carboxymethylhydroxyethyl-		
15		cellulose	0.3 - 0.6	0.3 - 2.3
		Alkoxyhydroxypropylhydroxy-		
	•	ethylcellulose	-	1.5 - 3.5
		Alkoxy group:		
20		straight-chain or branched,	2 - 8 carbon at	oms, 0.05 -
		50%, relative to the weight	of the substitute	ed cellulose
		ether		
		Carboxymethylated methyl-	DS OCH,	MS
25		hydroxyethyl- or	1.3 - 2.0	0.05 - 0.5
		methylhydroxypropylcellulose	1.3 - 2.2	0.1 - 1.0
٠	2.	Starch ethers	. 70	
		Hydroxypropylstarch:	MS 0.1 - 0.8	
		Carboxymethylstarch:	DS 0.1 - 0.8	
30		Hydroxypropylcarboxy-	•	
•		methylstarch:	DS 0.1 - 0.5/MS	0.1 - 0.8
•		Viscosity limits:	1 - 10,000 mPass	= (-2:3=aqueous=
			solution) measu	
			Höppler viscome	
35			and in distilled	

3. Guar gum derivatives

Guar endosperm flour (natural galactomannan) Viscosity (1% in water): 100-20,000 mPa.s (measured using a Brookfield RVT, 25°C 20 rpm)

Carboxymethylguar

Viscosity (3% in water):

100-20,000 mPa.s

10 Hydroxypropylguar

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Viscosity (1% in water):

100-10,000 mPa.s

Carboxymethylhydroxypropylguar

Viscosity (1% in water):

100-20,000 mPa.s

Cationized guar

(quaternary substitution

DS about 0.13)

Viscosity (1% in water):

100-20,000 mPa.s

4. Xanthan gum

(Polysaccharides, anionic, prepared by fermentation and extraction of

Xanthomonas campestris)
Viscosity (1% in water):

water): 100-10,000 mPa.s

- 5. Carobseed flour
- 6. Alginates
- b) Superabsorbents (SAPs)

Superabsorbents are lattice structures based on neutralized polyacrylic acid formed in free-radical copolymerization of monomers in the presence of small amounts of certain crosslinking agents.

Functi nality and reactivity of the cr sslinking agents is of great importance for th properties of a superabsorbent.

There are 2 main principles for preparing a polymer lattice structure:

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- a) the lattice (lattice structure) is synthesized in a single polymerization step, starting with the monomers, for which certain crosslinking agents are required for branching and formation of the lattice structure.
- b) A previously synthesized straight-chain or branched polymer is crosslinked.

For the preparation of SAPs, the method mentioned under a) has by far the greatest importance; for specific products, method b) is used. Various techniques are used for the polymerization process (for example emulsion polymerization and gel polymerization). In this process, an aqueous solution of the monomer acrylic acid is polymerized in the presence of small amounts of a cross-linking agent (polyfunctional monomer). Water serves as diluent and solvent (exothermic process) so as to avoid high temperatures. Polymerization can be carried out in a reactor or by means of the thin-film process; concentrations of up to 60% are customary.

25 The superabsorbents can also be grafted with starch. This grafting with starch is carried out during the polymerization and crosslinking process in a one-pot process. This starch is heated in water to dissolve it and is added as a solution at the beginning of the polymerization process. The proportion of starch, relative to the SAP, can be 3 - 30% by weight.

Superabsorbents can only absorb in the pH range from 4 to 10; (partial) neutralization can be carried out before or after polymerization. The product formed by

polym rizati n is a moist gel, which is comminuted using a gel cutter and then dried.

Cr sslinking for preparing SAPs substantially takes plac via covalent bonds, i.e. by using polyfunctional monomers during polymerization (crosslinking agent containing double bonds).

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Further crosslinking possibilities via covalent bonds are provided by polyfunctional molecules (diepoxides, azirines, polyalcohols). Already present, slightly crosslinked polymer chains are additionally crosslinked on the surface by means of the crosslinking agents mentioned.

This makes it possible to combine the good absorption properties of the slightly crosslinked superabsorbents (which have high absorption capacity but also large amounts of extractable material and a low absorption ratio, due to gel blocking) with the advantages of highly crosslinked SAPs (no gel blocking, better absorption capacity and absorption stability under compressive stress).

Important crosslinking reagents for the preparation of SAPs are

- methylenebisacrylamide; a crosslinking agent having high reactivity but leading to SAPs of less resistance to compression;
- b) diethylene glycol dialkyl ether; a crosslinking agent having less reactivity, which makes it possible to prepare SAP gels having better elastic properties and higher stability under compressive stress;
- c) vinyl compounds (vinyl ethers, vinyl esters, vinyl-phosphonic acid).

In order to achieve an optimum crosslinking and lattice structure of the SAPs, mixtures of crosslinking reagents

of high and low reactivity are used in the preparation process, it being possible to modify the density and constitution of the lattice structures over a wide range and adjust them as required by varying the mixing ratios of the crosslinking components used accordingly.

Suitable components c) are:

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- Polycondensation products of naphthalenesulfonic acid with formaldehyde in the form of their alkali metal salts, ammonium salts and alkaline earth metal salts;
- 2. Sulfo-modified melamine/formaldehyde polycondensation products in the form of their alkali metal salts or ammonium salts.
- The pulverulent and granulated components a), b) and if desired, c) can be individually admixed to the building material products or first components a), b) and c) are mixed by themselves, and this mixture is then—added to the building material products. Incorporation of these components in the building material products takes place by conventional stirring and mixing processes, for example by kneading. The sum of components a), b) and c) in the building material products is 0.01 to 2, preferably 0.1 to 0.5, % by weight, relative to the dry composition.
 - Components a), b) and c) can also be mixed with one another during their preparation, for example by thorough mixing or kneading before or after the drying process during preparation of the respective components. A subsequent joined fine milling produces particularly homogeneous and rapidly acting thickening systems. For combinations obtained by simply mixing the components, superabsorbents having a particle fineness of < 0.1 mm have proven to be particularly suitable.

The mixtures described have a strong thickening effect

in aqueous medium. In addition, d p nding on the particular composition of th se mixtures, some of them exhibit marked pseudoplasticity, which is mediated to liquidic/pasty systems and causes therein a remarkable structuring effect. The mixtures described are therefore particularly suitable as additives for building material products. These include paste-like tile adhesives, fillers and jointing compounds (in powder form or paste-like, in combination with plastic dispersions, polyvinyl alcohols, animal glue, casein as organic binder), jointing compounds based on polyacrylate and made elastic or plastic, plastic emulsion paints.

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This improvement in the structural properties in combination with an increased thickening effect caused by the combinations described makes it possible to omit completely or in part previously required mineral thickening and structuring aids, such as, for example, bentonites, attapulgites, highly disperse silicas. The substantial disadvantage of these inorganic thickeners, i.e. the strong increase in shrinkage behavior before and after hardening or drying of building material systems prepared therewith, can be reduced to a large extent or eliminated by the combinations described. Moreover, they lead to a reduction in costs as a result of reduced amounts used of the combinations compared with the customary amounts of natural polymers added. The claimed combinations moreover improve the processability and smoothness of liquidic pasty systems. The tackiness of building material products containing highly viscous cellulose ethers as thickeners and water retention agents, which is often troublesome, is significantly reduced by using the combinations -- mentioned. In mineral-based material systems toc, additions of the claimed combinations have a positive effect on processability. The resulting reduced tackiness enables the mortar and plastic compositions to be smoothened more easily. These properties also have an advantageous effect in concrete mixtures (underwater concrete, gunned concrete, pumped

concrete), whose homogen ity, stability and pumpability are improved.

Examples:

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For the mixtures described below, the following individual components were used:

	Viscosity level mea- sured in 21 aqueous solution	Degree of ether catio	ifi-
Methylhydroxyethyl- cellulose (1) Methylhydroxyethyl-	30,000	1.5	0.12
cellulose (2) Methylhydroxyethyl-	50,000	1.5	0.11
cellulose (3) Methylhydroxyethyl-	6,000	1.5	0.13
Cellulose (4) Methylhydroxyethyl-	6,000	1.5	0.10
Cellulose (5) Methylhydroxyethyl-	15,000	1.8	0.12
Cellulose (6) Methylhydroxyethyl-	6,000	1.6	0.15
Cellulose (7) Methylhydroxyethyl-	30,000	1.55	0.21
Methylhydroxyethyl-	6,000	1.64	0.24
Cellulose (9)	4,000	1.88	0.21
Hydroxyethylcellulose Carboxymethylhydroxy-	6,000	-	2.1
ethylcellulose	3,000	0.55	0.75
Ethylhydroxyethylcellulose Methylhydroxypropyl-	1,000	0.97	1.9
cellulose	4,000	1.88	0.21
Alkoxyhydroxypropylhydroxy- ethylcellulose	25,000	DS=OC,H,(()	OH) -
		MS=OC,H. 2.45	
Alkoxyhydroxypropylhydroxy- ethylcellulose	25,000	DS=OC,H,(C-OC,H,	
ar .		MS = OC,B,	

5		Viscosity level mea- sured in 2% aqueous solution	Degree of eth catio	rifi-
10	Hydropropylstarch (crosslinked with epichlorohydrin)	1,300 (5% aqueous solution)	-	0.64
	Guar gum (Mepro guar CSA 200/50)	5,000 (1% solution)	_	-
15	Hydroxypropylguam (Jaguar ^(a) 8060)	4,000 (1% solution)	-	0.42
20	Carboxymethylhydroxy- propylguam (Jaguar ^(R) 8600)	3,800 (1% solution)	0.06	0.40
25	Guar gum, quaternary- substituted (Meyprofloc 130)	2,000 (1% solution)	degree quate subst	ernary itution
30	(Kilodopot,20 MD)	starting viscosity: 6300 mPa.s (2% solution)		

	sorpti capaci deion-	ty in	Centri- fuging reten- tion 0.9% NaCl sol. (g/g)	Absorp- tion capacity under pressure 0.9% NaCl sol. (ml/g)	Extractable material
Superabsor- bent (1) Sanwet ^(R) IM 1500	500	50	32	15	7
Superabsor- bent (2) Sanwet ^(R) IM 1000	1000	65	42	5	14
Superabsor- bent (3) Sanwet ^(R) IM 5000s	400	48	32	32	3

Naphthalenesulfonic acid/formaldehyde condensation product;

Sodium salt:

5	Properties	Typical data
-	Supplied	in liquid/pulverulent form
	Concentration (active conten	t)20 - 40%/70 - 96%
	sodium sulfate content	0 - 25%
1.0	Molecular weight	4,000 - 40,000
10	рH	6.5 - 11
	Viscosity (23°C)	10 - 150 mPa.s liquid
	•	products)

Melamine/formaldehyde condensation product, sulfonated; sodium salt:

15	Properties	Typical data
	Supplied	in liquid/pulverulent form,
20	Concentration (active content Molecular weight pH	20,000 - 30,000 8 - 12
	Density	liquid products: 1.1 g/cm ³ solid products: 1.7 - 1.9 g/cm ³

The mixtures described below of the abovementioned individual components were prepared by simply mixing the components in the relative amounts given. In each case, pbw is parts by weight.

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5			Viscosity of the combination described, measured as an X% aqueous solution mPa.s
10	1	<pre>1 pbw of methylhydroxy-</pre>	30,500 (as 1.3% aqueous solution)
15 20	2	1.7 pbw of methylhydroxy- ethylcellulose (2) + 0.21 pbw of superabsorbent (2) + 0.009 pbw of naphthalene- sulfonic acid/formal- dehyde condensation product sodium salt	145,000 (as 2% aqueous solution)
25	3	1.4 pbw of methylhydroxy- ethylcellulose (3) + 0.40 pbw of superabsorbent (1) + 0.20 pbw of naphthalene- sulfonic acid/formal- dehyde condensation product sodium salt	109,000 (as 2% aqueous solution)
35	4	1.6 pbw of methylhydroxy- ethylcellulose (4) + 0.2 pbw of superabsorbent (1) + 0.2 pbw of naphthalene- sulfonic acid/formal- dehyde condensation product sodium salt	88,000 (as 2% aqueous (solution)
40	5	1.6 pbw of methylhydroxy- ethylcellulose (5) + 0.3 pbw of superabsorbent (2) + 0.1 pbw of naphthalene- sulfonic acid/formal- dehyde condensation product sodium salt	102,000 (as 2% aqueous solution)
43 '	6	1.8 pbw of methylhydroxy- ethylcellulose (6) + 0.2 pbw of superabsorbent (1)	73,000 (as 2% aqueous solution)
50	7	0.5 pbw of methylhydroxy- ethylcellulose (7) + 0.5 pbw of superabsorbent (1)	27,500 (as 1t aqueous solution)
	_		

		*	Viscosity of th combination described, measured as an X% aqueous solution mPa.s
10	8 1.4 pbw + 0.4 pbw + 0.2 pbw	of hydroxyethyl- cellulose of superabsorbent (1) of naphthalene- sulfonic acid/formal- dehyde condensation product sodium salt	89,500 (as 2% aqueous solution)
15	9 1.8 pbw + 0.2 pbw	of hydroxyethyl- cellulose of superabsorbent (1)	30,000 (as 2% ageuous solution)
20	10 1.6 pbw + 0.2 pbw + 0.2 pbw	of hydroxyethyl- cellulose of superabsorbent (1) of naphthalene- sulfonic acid/formal- dehyde condensation product sodium salt	48,000 (as 2% aqueous solution)
· 25 · · · · · · · · · · · · · · · · · · ·	11 1.4 pbw + 0.4 pbw + 0.2 pbw	of carboxymethyl- hydroxyethylcellulose of superabsorbent (1) of naphthalene- sulfonic acid/formal- dehyde condensation product sodium salt	7,500 (as 2% aqueous solution)
35	12 1.7 pbw + 0.3 pbw		14,500 (as 2% aqueous solution)
40	13 1.4 pbw + 0.4 pbw 0.2 pbw		solution)
45 .: 50	14 1.4 pbw + 0.4 pbw + 0.2 pbw	of methylhydroxy- propylcellulose of superabsorbent (1) of naphthalene- sulfonic acid/formal- dehyde condensation product sodium salt	solution)

5			Viscosity of the combinati n described, measured as an X% aqueous solution mPa.s
10	15 1.4 pbw + 0.4 pbw + 0.2 pbw	of naphthalene- sulfonic acid/formal- dehyde condensation	43,000 (as 2% aqueous solution)
15	16 1.7 pbw + 0.3 pbw	product sodium salt of alkoxyhydroxy- propylhydroxyethyl- cellulose (1) superabsorbent	78,000 (as 2% aqueous solution)
20 25	17 1.7 pbw + 0.2 pbw + 0.1 pbw	of alkoxyhydroxy- propylhydroxyethyl- cellulose superabsorbent (1) naphthalenesulfonic- acid/formaldehyde condensation product sodium salt	80,000 (as 2% aqueous solution) solution)
30	18 1.2 pbw + 0.8 pbw	of methylhydroxy- ethylcellulose (8) of superabsorbent (3)	450,000 (as 2% ageuous solution)
35	19 1.7 pbw 0.3 pbw	of hydroxypropyl- starch (crosslinked with epichlorohydrin) of superabsorbent (1)	13,500 (as 2% aqueous solution)
	20 1.7 pbw + 0.3 pbw	of guar gum (guar endosperm flour) of superabsorbent (3)	94,000 (as 2% aqueous solution)
40	21 1.7 pbw + 0.3 pbw	of hydroxypropylguar of superabsorbent (3)	61,000 (as 2% ageuous solution)
45	22 1.7 pbw 0.3 pbw	hydroxypropylguar	44,000 (as 2% aqueous solution)
50 55	23 1.7 pbw 0.2 pbw + 0.1 pbw	of superabsorbent (3)	solution)

5		Viscosity of th combination described, measured as an X% aqueous solution mPa.s
	24 1.7 pbw of guar gum, quater- nary-substituted + 0.3 pbw of superabsorbent (3)	measurable agreeme
10	25 1.4 pbw of methylhydroxy- ethylcellulose (3) + 0.4 pbw of superabsorbent (3) + 0.2 pbw melamine/formaldehyde condensation product, sulfonated, sodium salt	
20	1.4 pbw of hydroxyethyl- cellulose + 0.4 pbw of superabsorbent (3) + 0.2 pbw of melamine/formal- dehyde condensation product, sulfonated, sodium salt	32,000 (as 2% ageuous solution)
· · ` 30	27 1.4 pbw of carboxymethyl- hydroxyethylcellulose + 0.4 pbw of superabsorbent (2) + 0.2 pbw of melamine/formal- dehyde condensation product, sulfonated, sodium salt	4,500 (as 2% aqueous solution)
35 . 40	propylcellulose	39,000 (as 2% ageuous solution)
45.	+ 0.2 pbw of melamine/formal- dehyde condensation product, sulfonated, sodium salt	14,300 (as 2% aqueous solution)
50 3	+ 0.2 Phy of superabsorbent (1) (5,700 as 2% aqueous olution)

5			Viscosity of the combinati ndescribed, measured as an Xt aqueous solution mPa.s
10	+ 0.8 pbw of sur + 0.2 pbw of mel dehyde	erabsorbent (1) amine/formal- condensation t, sulfonated,	7,400 (as 2% aqueous solution)
15 20	+ 0.4 pbw of sup + 0.2 pbw of nap sulfor dehyde	hylhydroxyethyl ose (9) erabsorbent (3) hthalene- ic acid/formal- condensation t, calcium salt	102,000 (as 2% ageuous solution)
25	cellul + 0.4 pbw of sup + 0.2 pbw of nap sulfon dehyde	roxyethyl- ose erabsorbent (3) hthalene- ic acid/formal- condensation t, calcium salt	63,500 (as 2% aqueous solution)
30 35	+ 0.4 pbw of sup + 0.2 pbw of nap sulfon dehyde	r gum (guar erm flour) erabsorbent (3) hthalene- ic acid/formal- condensation t, calcium salt	70,000 (as 2% aqueous solution)
40	+ 0.4 pbw of sup + 0.2 pbw of nap sulfon dehyde	erabsorbent (3)	55,800 (as 2% aqueous solution)
45 -	+ 0.4 pbw of sup + 0.2 pbw of nap sulfon dehyde	erabsorbent (3)	14,600 (as 2% aqueous solution)

Viscosity measurements of superabsorbent in water or superabsorbent + polycondensation products without use of derivatives of natural polymers

	đ	q	S	Ð	60	44	Б	Я		+	K
	₽dd	Mdd	₽ďď	₽₽₩	∌dd	₽ďď	add	Mdd	Add	ppw	₽ďď
Superabsorbent (1)	0.2	0.2	0.2	0.4	••0	0.4	9.0	9.0	9.0	0.4	0.4
Maphthalenesulfonic acid/formaldehyde polycondensation product, sodium salt z calcium salt	e e	0.2		•	0.2	♦.0	1	0.2	1	0.1	
Melamine/formal- d hyde polycondensa- tion product, sulfon- ated, sodium salt	•	•	0.2	1	1	0.2			0.2	1	0.1
Water	99.8	9.66	99.6	9.66	1.66	₹66	₽.66	99.2	99.2	99.5	99.5
Viscosity measured using Brookfield RV 20 rpm, 25°C RT	3200	09	55	12000	200	185	20000	000	0000	1500	1400
Not s on appearance of samples:	gel- like mater- ial non- flow-	yel- low- ish, clear, flow- able	clear, color- less flow- able	gel- like mater- non- flow- able	yel- low- ish, clear, flow- able	clear, color- less flow- able	gel- like mater- ial non- flow-	yel- low- ish, clear	clear, color- less	yel- low- ish, clear, flow-	clear, color- less, flow-

Results of the viscosity masurements (Brookfield RV, 20 rpm, 25°C Rt)

No.	No. Combinations, comprising:	Viscosity of methyl-hydroxyethylcellulose as 1% aqueous solution mPa.s	Viscosity of combination described, measured as X% aqueous solution mPa.s
đ	1.7 pbw of methylhydroxyethylcellulose + 0.3 pbw of superabsorbent (1)		29,300 mPa.s (as 1% aquecus solution)
q	1.7 pbw of methylhydroxyethylcellulose + 0.2 pbw of superabsorbent (1) + 0.2 pbw of naphthalenesulfonic acid/ formaldehyde condensation product sodium salt		26,000 mPa.s (as 1% aqueous solution)
U	methylhydroxyethylcellulose without additives	2,400	•

Examples of Calding material products cording to the invention

1. Paste-like tile adhesive

1		, , , , , , , , , , , , , , , , , , , 	·		, , , , , , , , , , , , , , , , , , ,
		a (compar- ative sample)	b	C	d -
:	Components	pbw	pbw	pbw	pbw
	Filler mixture comprising finely ground calcium carbonate and ground mica	67	67	67	67
	Bentonite	1	1	. 1	0.8
	Hydroxyethylcellulose, viscosity level 100,000 mPa.s (2% ageuous solution)	0.68	0.4	0.30	0.30
	Superabsorbent (1)	-	0.2	0.25	0.30
	Polycondensation product based on naphthalene-sulfonic acid/formaldehyde alkali metal salt	-	-	0.05	
	Water	20-25	20-25	20-25	20-25
	Plastic dispersion (based on styrene/acrylate)	10	10	10	10
	Antifoam	0.1	0.1	0.1	0.1
	Preservative	0.15	0.15	0.15	0.15
	Aminomethylpropanol (as alkalising agent)	0.1	0.1	0.1	0.1
	Antisliding behavior of stonework_tiles (10x10cm) on concrete (mm)	2-3	1-2	2-3	0
	Skinning time (min) tested with stonework tiles (5x5 cm) on concrete Tiles loaded with 1 kg Tiles loaded with 2 kg	20 30	25 35	25-30 35-40	20-25 30-35

2. R ady-mixed jointing compound

	a	b .	C	d
Filler mixture (fin ly ground calcium carbonate 0 - 20 µm)	58	58	58	58
Fin ly ground mica	4	4	4	4
Attapulgite, finely ground (as inorganic thickener)	2.5	-	2.5	-
M thylhydroxyethyl- c llulose or methyl- hydroxypropylcellulose visc. l v l 15000-60000 mPa.s (2% aqueous solution)	0.50	0.25	0.40	0.30
Hydroxyethylcellulose visc. level 15000-60000 mPa.s (2% aqueous s lution)	•	·	0.10	•
Sup rabsorbent (1)	-	0.25	-	0.20
Water	30-35	30-35	30-35	30-35
Plastic dispersion (con- taining polyvinyl acetate plasticized with 10% of dibutyl phthalate	4-6	4-6	4-6	4-6
Antifoam	0.1	0.1	0.1	0.1
Preservative	0.1	0.1	0.1	0.1
Pr pylene glycol	1.0	1.0	1.0	1.0
Consistency of the compound and processability	somewhat viscous and tacky, exten- sive after- thicken- ing	proces- sable no after- thick- ening	more easily process- sable than less tacky, exten- sive after- thick- ening of the paste- like mater- ial	easily proc- cess- able, a,no after- thick- ening of the paste- like mat- erial

Continuation of the Table: R ady-mix jointing compound

	a	b	С	d
Shrinkage behavior of th hardened mat rial on gypsum-board sheeting	xten- siv shrink- age, crack- ing in the hard- ened mater- ial	r du- c d shrink- age, hardly any cracks	xt n- siv shrink- age, too many cracks in the hard- ened mater- ial	no shrink age, hardly any crack- ing in the hard- ened mater- ial

3. Pulverulent jointing compound (without addition of gypsum)

	a .	ь	С	d
Piller mixture (compris- ing finely ground calcium carbonate, mica, talc)	92-94	92-94	92-94	92-94
Bentonites	0.6-1	0.6-1	0.5	- .
Plastic dispersion powder (copolymer of vinyl acetate/ethylene or versatic ester)	3-6	3-6	3-6	3-6
Polyvinyl alcohol (low- molecular-weight)	0.4-0.8	.4-0.8	0.4-0.8	0.4-0.8
Methylhydroxyethyl- cellulose or methyl- hydroxypropylcellulose visc. level 10000-100000 mPa.s (2% aqueous solution)	0.60	0.24	0.20	0.35
Carboxymethylhydroxyethyl- cellulose visc. level 3000-10000 mPa.s (2% aqueous solution)	-	0.30	0.30	-
Superabsorbent (1)	-	0.06	0.1	0.25
Addition of water per 100 g of pulverulent mixture	45-50	45-50	45-50	45-50
Consistency and process- ability of the compound mixed with water	good, slight struc- turing	good, slight impro- vement in the struc- ture	very good dis- tinct impro- vement in the struc- ture of the com- pound	very good, good struc- ture and stab- ility

5.

Continuation Table Pulverulent jointing compound

•	a	b	С	d
Stability of consistency of the compound mixed with water	gradu- ally after thick- ening (after 7 days)	hardly any after thick- ening	no after thick- ening	no after thick- ening
Shrinkage behavior and cracking on gypsum-board sheeting	exten- sive crack- ing	less crack- ing	hardly any shrink- age hardly any crack- ing	hardly any shrink- age no crack- ing

The use of the polycondensation products in the combinations described of d rivatives of natural polym rs with sup rabsorbents leads to transpar nt, predominantly still pourable polymer solutions. Particularly advantageous are additions of melamine/formaldehyde condensation products, resulting in completely transparent and colorless solutions.

Solutions of natural polymers show in many cases naturally more or less extensive clouding, which is removed by addition of superabsorbents and melamine/formaldehyde condensation products. This optical effect represents an improvement in the quality of natural polymers in those applications in which solubility of the polymers in water with the formation of a transparent solution is necessary.

HOR 91/P 086

THE CLAIMS DEFINING THE INVENTION ARE AS FOLLOWS:

- 1. A building material product comprising a mixture of
- a) at least 10% by weight, in particular 10 to 99% by weight, of a water-soluble natural polymer and derivatives thereof,
- b) 1 to 89% by weight of an alkali metal salt or ammonium salt of a crosslinked polyacrylate which, if desired, is grafted with a starch, and
- o to 60% by weight of a water-soluble alkali metal salt or alkaline earth metal salt of arylsulfonic acid/formaldehyde condensation products or of a sulfo-modified melamine/formaldehyde polycondensation product.
- 2. A building material product as claimed in claim 1, comprising a mixture of 40 to 90% by weight of component a), 5 to 60% by weight of component b) and 1 to 40% by weight of component c).
- 3. A building material product as claimed in claim 1, comprising 0.01 to 2, preferably 0.1 to 0.5 % by weight, relative to the dry composition, of the mixture of components a), b) and c).

DATED this 18th day of March 1992.

HOECHST AKTIENGESELLSCHAFT

WATERMARK PATENT & TRADEMARK ATTORNEYS "THE ATRIUM"
290 BURWOOD ROAD
HAWTHORN. VIC. 3122.

Search Title: 00-2040 opt User: cpabuc - Chris Burton, N2 55
PAN: 94-188748, Page 1 of Tue Sep 19 18:32:04, VIEWED MERKED

*KAOS F64 94-188748/23 *JP06127892-A Cement compan, for extrusion moulding for mfg. outer walls of houses, etc. comprises powder of cellulose and high performance water reducing agent

KAO CORP 92.10.16 92JP-278579

A93 L02 (A21) (94.05.10) COHE 24/00, B28B 3/20, COHE 24/38, 28/02

Cement compsn. comprises a powder of cellulose and high-performance water-reducing agent; The mean grain size of the cellulose is 2-100 microns and the water-reducing agent is e.g. a formaldehyde condensate of naphthalene sulphonate.

USE - Used for making outer walls, roofs or floors of houses. (4pp

Dwg.No.0/0)

N94-148808

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